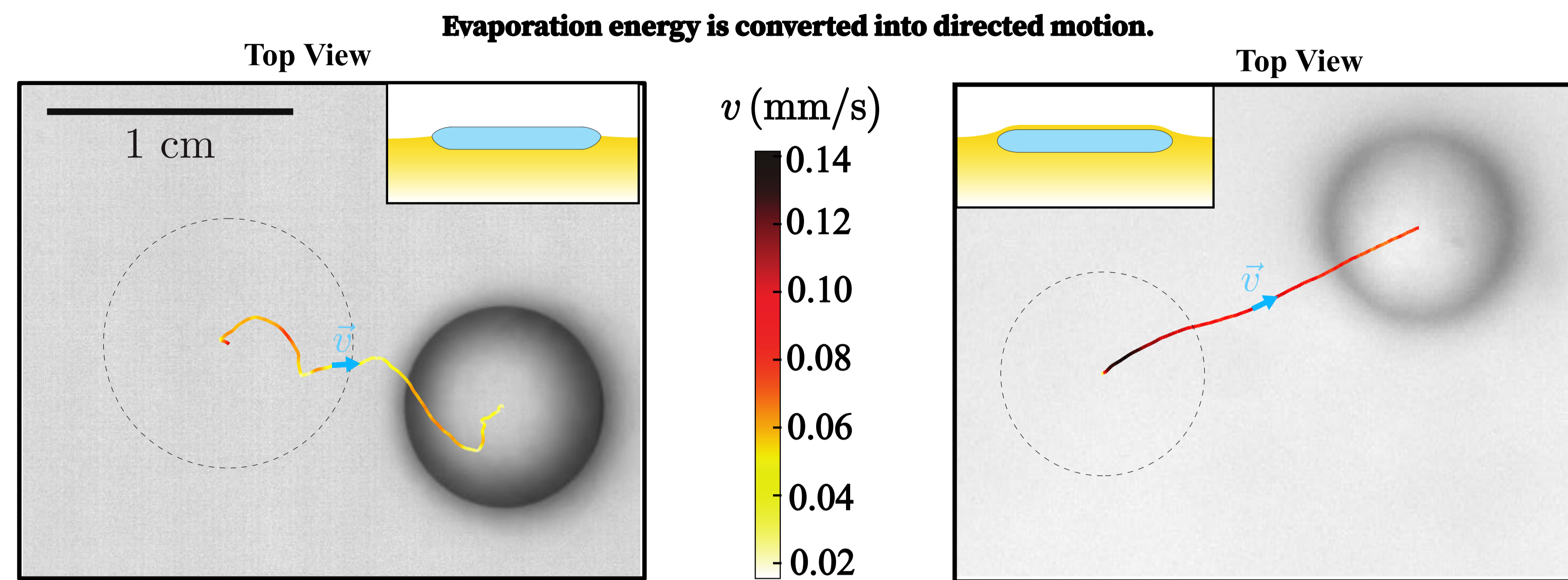


Active Volatile Drops on Liquid Baths

Benjamin Reichert, Jean-Benoît Le Cam, Arnaud Saint-Jalmes and Giuseppe Pucci
Soft Matter Group, Institute of Physics of Rennes UMR 6251, Rennes, France

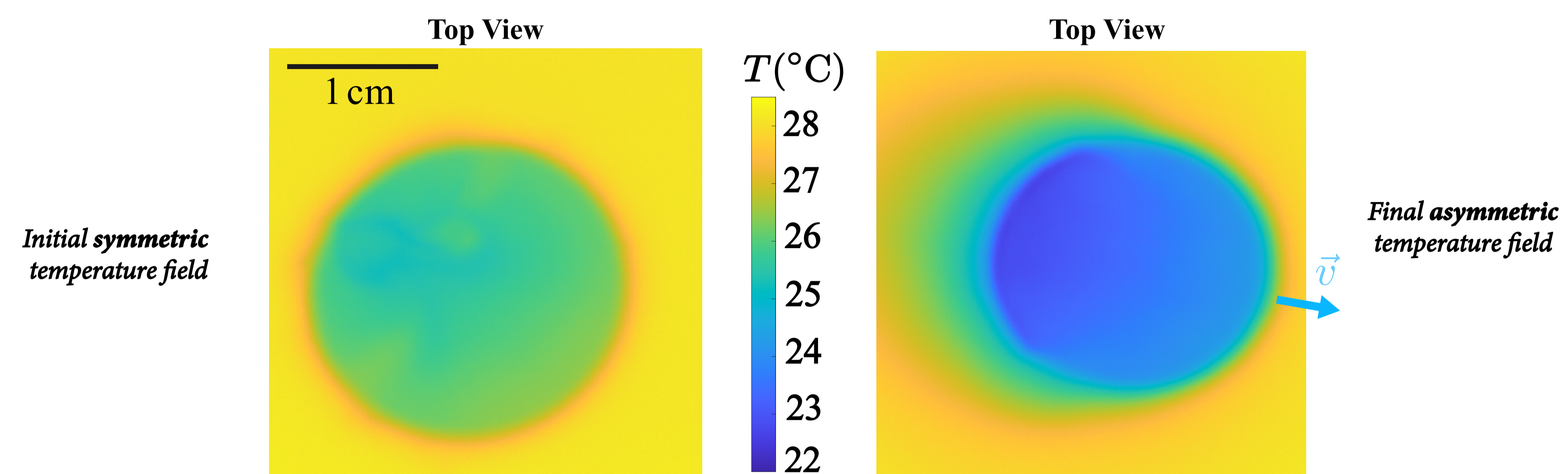
When a drop of volatile alcohol is deposited onto the surface of a bath of immiscible liquid, the drop spontaneously propels on the surface.



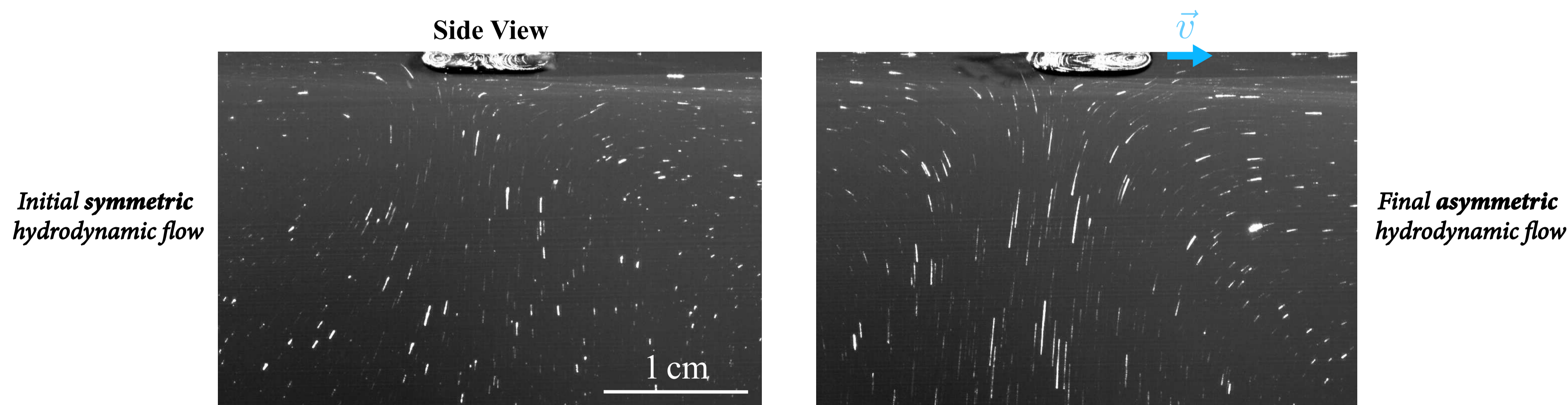
The presence of a thin film of bath liquid coating the drop is associated to straighter trajectories.

Self-propulsion is triggered by a thermocapillary convective instability.

Spontaneous symmetry breaking of the surface temperature field...



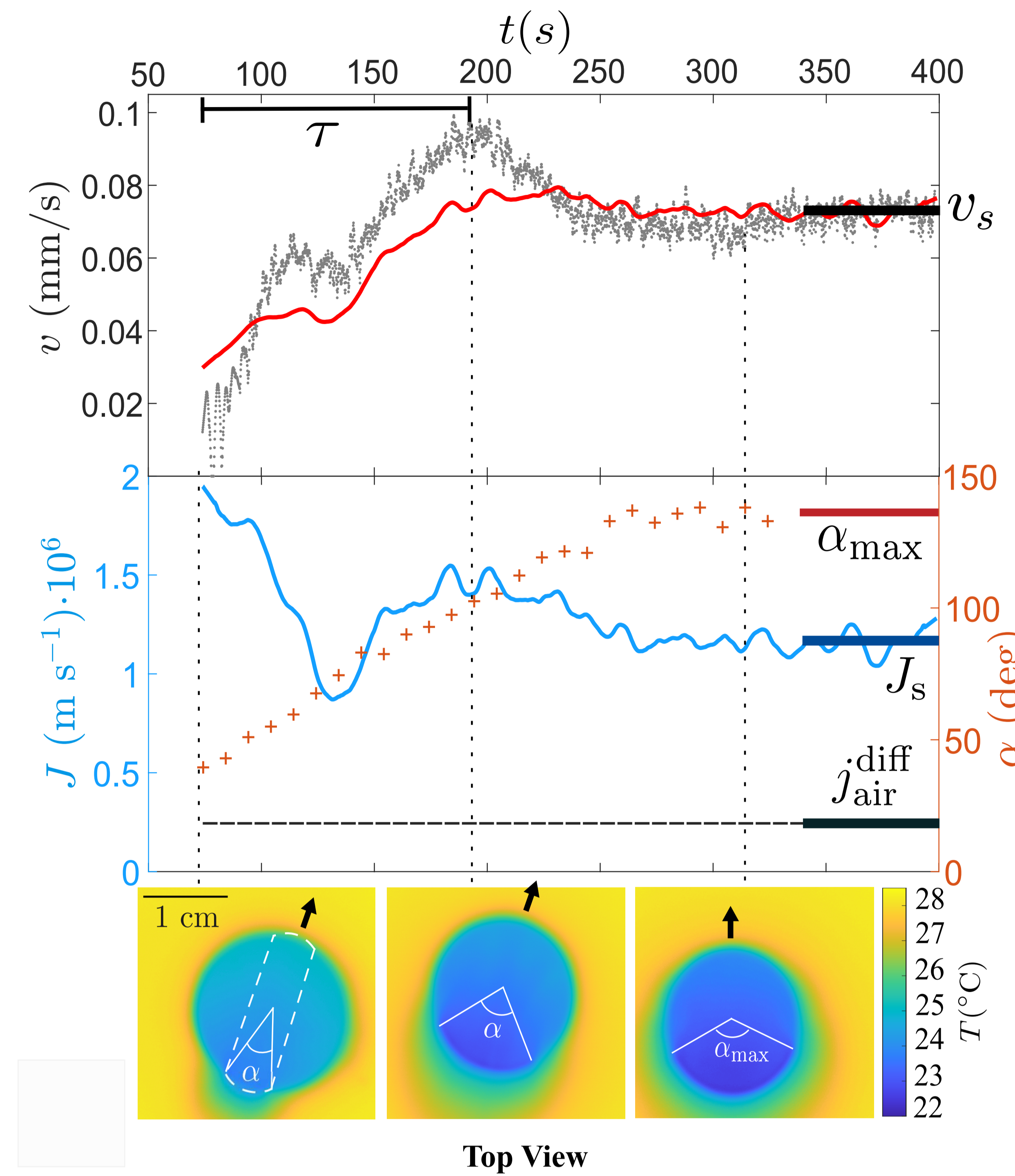
...results into asymmetric thermocapillary stresses and symmetry breaking of the hydrodynamic flows.



A propulsive force emerges as a result of the viscous stress response of the liquid bath to the Marangoni stress exerted on the drop's lower interface.

In contrast to a solid Marangoni surfer, our drop propels in a direction opposite to the interfacial tension gradient.
The propulsion scheme is rather similar to a classical squirmer developing tangential stress on its interface to sustain its locomotion.

Relation between the drop velocity and the evaporation flux J (i.e. the activity source) in the transient regime?



Force balance between Marangoni traction and Stokes drag

$$v \sim \left| \frac{d\gamma}{dT} \right| \frac{\Delta T}{\eta_2} \quad \text{with } \Delta T = T^{++} - T^+$$

Interfacial tension γ
Bath viscosity η_2

Thermal balance dominated by heat convection

$$\rho_2 C_p \Delta T u_M \propto \rho_1 \mathcal{L}_v J$$

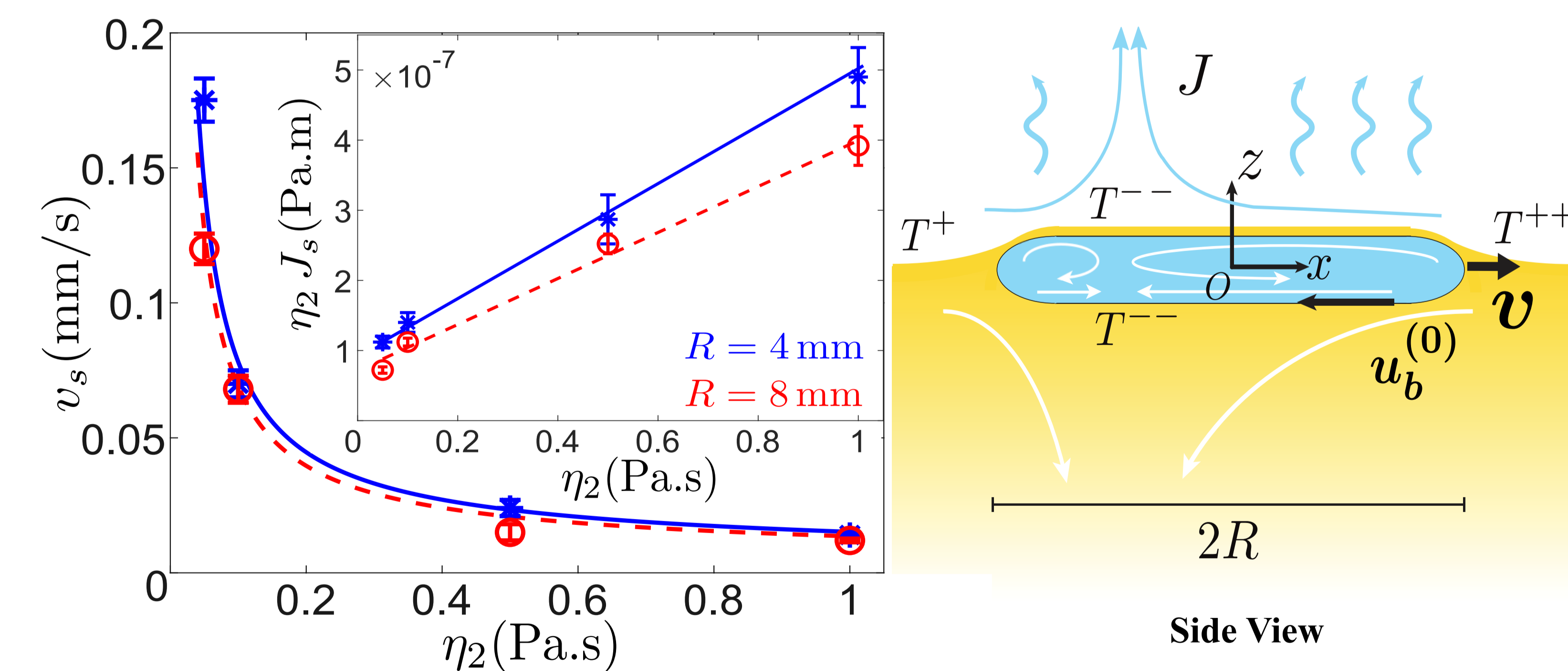
with the velocity of the Marangoni flow

$$u_M \sim \left| \frac{d\gamma}{dT} \right| \frac{\Delta T}{\eta_2}$$

Combining force and thermal balance

$$v \sim \left[\frac{\rho_1 |d\gamma/dT| \mathcal{L}_v J}{\rho_2 \eta_2 C_p} \right]^{1/2}$$

In the stationary regime, the activity of the system can be tuned by varying bath viscosity η_2



Evaporation is governed by two transport processes of vapour in air, diffusion and convection.

The evaporation flux is the sum of a diffusion and a convection contribution

$$J_s = a + b/\eta_2$$

Drop velocity at variable activity is well captured for different radii by combining force and thermal balances

$$v_s \sim \left[\frac{\rho_1 |d\gamma/dT| \mathcal{L}_v (a + b/\eta_2)}{\rho_2 \eta_2 C_p} \right]^{1/2}$$

